

RESOLUTION AND ABSOLUTE CONFIGURATION OF TRANS (+)- AND (-)-6-HYDROXYMETHYL-2-METHOXY-5,6-DIHYDRO-2H-PYRANS, SUBSTRATES FOR TOTAL SYNTHESIS OF MONOSACCHARIDES†

A. KONOWAŁ, J. JURCZAK and A. ZAMOJSKI*

Institute of Organic Chemistry, Polish Academy of Sciences, 01-224 Warsaw, Poland

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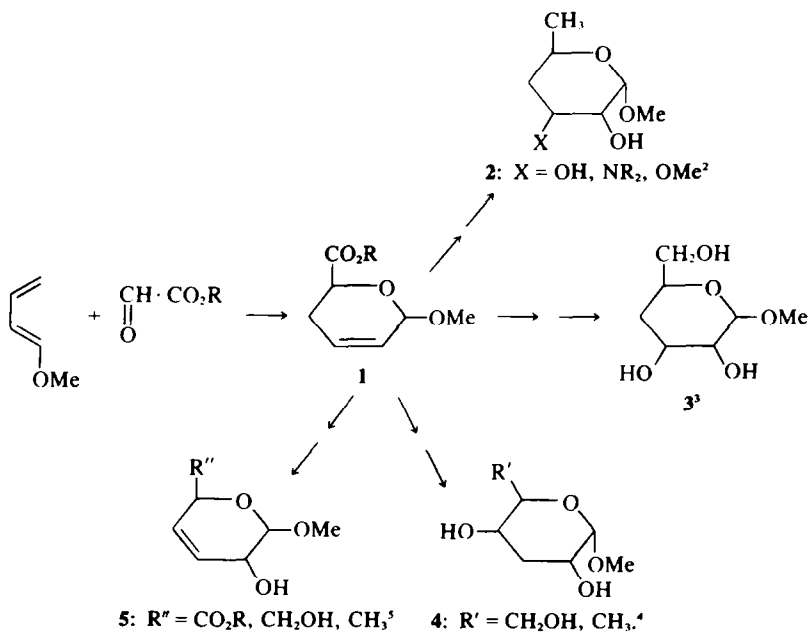
Abstract—*trans* 6-Hydroxymethyl-2-methoxy-5,6-dihydro-2H-pyran was resolved into enantiomers by means of ω -camphanic acid ester. The 2*S*:6*S* configuration was determined for the laevo rotating enantiomer.

It has been shown¹ that easily available esters of 2-methoxy-5,6-dihydro-2H-pyran-6-carboxylic acid (1) can be converted in a few, remarkably stereoselective steps into a variety of monosaccharides.

Thus, the reactions shown in Scheme 1 represent a general, totally synthetic approach to monosaccharides. Methyl glycosides 2-5 have been obtained as pure stereoisomers in racemic form.

glyoxylates.⁸ This approach failed because of low enantiomeric purities of the adducts obtained.

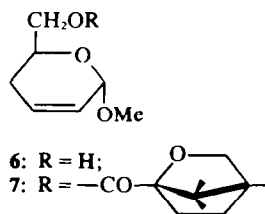
We decided therefore to resolve the key substrate of the hexose synthesis into enantiomers: *trans*-6-hydroxymethyl-2-methoxy-5,6-dihydro-2H-pyran⁹ (6) which is easily obtainable by LAH reduction of 1. From the variety of methods used for resolution of racemic alcohols we chose the separation by fractional crystalliza-



Scheme 1.

As sugars occur in nature as enantiomers, valuable information regarding their configuration⁶ and conformation⁷ can be derived from optical rotation data. Consequently, the introduction of enantiomeric substrates should greatly enhance the usefulness of the approach shown in Scheme 1.

Our initial efforts were directed towards the synthesis of enantiomeric esters 1 by "asymmetric" condensation of 1-alkoxy-1,3-butadienes with optically active alkyl



tion of diastereoisomeric esters of 6 with (-) ω -camphanic acid. This new method, introduced by Gerlach,¹⁰ proved to be very efficient in preparative resolution

†Dedicated to Prof. Dr. Vlado Prelog on the occasion of his 70th birthday.

trated to dryness, and the residue was crystallized from a mixture of ether and *n*-hexane (7:3 v/v) at -15° . The crystals obtained ($[\alpha]_{D}^{25} + 18.8$ (c, 2.12)) were combined with the residues after the crystallization of ester (-)-7. The ^1H NMR spectrum ($\text{Eu}(\text{fod})_3$) indicated the proportion of diastereoisomeric esters 7 as 27:73. Two consecutive crystallizations from the same solvent gave the second pure diastereoisomeric ester (+)-7, 1.65 g, m.p. $72-75^{\circ}$, $[\alpha]_{D}^{25} + 26.2^{\circ}$ (c, 1.8). Its homogeneity was confirmed by the ^1H NMR spectrum ($\text{Eu}(\text{fod})_3$). The total yield of camphanic acid esters 7 (pure diastereoisomers and mixed fractions): was 13.6 g (69.4%). Analysis (for mixture of diastereoisomeric 7): (Found: C, 63.0; H, 7.4. Calc. for $\text{C}_{17}\text{H}_{24}\text{O}_6$: C, 62.9; H, 7.5%).

An attempt to separate diastereoisomeric 7 on a silica gel column gave only partially resolved fractions.

Hydrolysis of esters (-)-7 and (+)-7

6S - Hydroxymethyl - 2S - methoxy - 5,6 - dihydro - 2H - pyran [(+)-6]. Ester (-)-7 (1 g; 3.1 mmol, $[\alpha]_{D}^{25} - 29.8^{\circ}$) was suspended in 60 ml of 0.5 N KOH soln in water-EtOH (1:3) and refluxed for 1 hr. The EtOH was evaporated and the residue was extracted with EtOAc. The extract was washed with water, dried (MgSO_4) and evaporated to dryness. The remaining oil was distilled at $65-67^{\circ}/0.4$ to give 0.332 g (75%) of (-)-6 $[\alpha]_{D}^{25} - 80.0 \pm 0.5^{\circ}$ (c, 2.0).

6R - Hydroxymethyl - 2R - methoxy - 5,6 - dihydro - 2H - pyran [(+)-6] was obtained from ester (+)-7 in an analogous way. From 0.8 g of (+)-7 (2.5 mmol, $[\alpha]_{D}^{25} + 26.2^{\circ}$), 0.225 g (77%) of (+)-6 was obtained, b.p. $64-66^{\circ}/0.4$, $[\alpha]_{D}^{25} + 79.2 \pm 0.5^{\circ}$ (c, 1.97). The analytical and spectral data (IR and ^1H NMR) of (-)-6 and (+)-6 were identical with those of the racemic compound.⁹

Hydrogenation of enantiomeric 6 - hydroxymethyl - 2 - methoxy - 5,6 - dihydro - 2H - pyrans (-)-6 and (+)-6

Alcohols 6 were hydrogenated in methanolic soln under atmospheric pressure in the presence of 10% by weight of Adams platinum catalyst until absorption of hydrogen stopped. Compound (+)-8 was obtained from (-)-6 as an oil, b.p. $55^{\circ}/0.4$, $[\alpha]_{D}^{25} + 136.7 \pm 1^{\circ}$ (c, 2.06). Compound (-)-8 was obtained from (+)-6 as an oil, b.p. $55^{\circ}/0.4$, $[\alpha]_{D}^{25} - 140.7 \pm 1^{\circ}$ (c, 2.06). The analytical and spectral data (IR and ^1H NMR) of (+)-8 and (-)-8 were identical with those of the racemic compound 8.⁹ In the ^1H NMR spectra of both compounds, (-)-8 and (+)-8, signals of anomeric proton appeared on addition of ca. 17% of tris (3 - trifluoromethylhydroxymethyl)ethylene - d - camphorato) europium as singlets. In the spectrum of racemic 8 made under similar conditions the signal of an anomeric proton appeared as two signals separated by about 10 Hz.

Epoxidation of 6S - camphanyloxymethyl - 2S - methoxy - 5,6 - dihydro - 2H - pyrans [(+)-7]

Methyl 2,3 - anhydro - 6 - O - camphanyl - 4 - deoxy - α - D - lyxo - and α - D - ribo - hexopyranosides (9 and 10). To a soln of (-)-7 (3 g; 9.7 mmol) in 150 ml CHCl_3 , 80% *m*-chloroperbenzoic acid (3.7 g; 21.6 mmol) was added and the mixture was left at room temp. for 14 days. Progress of the reaction was followed by TLC in benzene-acetone 8:2. *m*-Chlorobenzoic acid was filtered off, chloroform soln was washed with 10% Na_2CO_3 aq. and thereafter with water. The soln was dried (MgSO_4) and evaporated to dryness. The residue was chromatographed on a silica gel (150 g) column with a mixture of ligroin 60-80 $^{\circ}$ and EtOAc 65:35 (v/v). The first fraction contained compd 9, 1.37 g (43.5%), colourless crystals, m.p. $98-100^{\circ}$ (from MeOH); $[\alpha]_{D}^{25} + 27.3^{\circ}$; $[\alpha]_{D}^{25} + 30.9^{\circ}$; $[\alpha]_{D}^{25} + 48.1^{\circ}$ and $[\alpha]_{D}^{25} + 65.8^{\circ}$ (c, 1.62 in CHCl_3). (Found: C, 60.1; H, 7.2. Calc. for $\text{C}_{17}\text{H}_{24}\text{O}_7$: C, 60.0; H, 7.1%). IR (KBr): 1795 cm^{-1} (lactone) 1730 cm^{-1} (ester) 1250, 915 and 840 cm^{-1} (epoxide); ^1H NMR (CDCl_3): 5.05 (1H, s) H_1 ; 3.55 (3H, s) OCH_3 ; 3.50 (1H, m) H_3 ; 3.10 (1H, d, $J_{2,3} = 4\text{ Hz}$) H_2 ; 1.16, 1.10 and 1.03 (9H, 3s) 3CH_3 .

The second fraction contained compd 10; 1.48 g (47%) colourless crystals, m.p. $143-145^{\circ}$ (from MeOH); $[\alpha]_{D}^{25} + 35.4^{\circ}$; $[\alpha]_{D}^{25} +$

40.1° ; $[\alpha]_{D}^{25} + 65^{\circ}$ and $[\alpha]_{D}^{25} + 93.5^{\circ}$ (c, 1.1 in CHCl_3). (Found: C, 59.8; H, 7.2. Calc. for $\text{C}_{17}\text{H}_{24}\text{O}_7$: C, 60.0; H, 7.1%). IR (KBr): 1790 cm^{-1} (lactone) 1740 cm^{-1} (ester) 1250, 915 and 855 cm^{-1} (epoxide); ^1H NMR (CDCl_3): 5.06 (1H, d, $J_{1,2} = 2.6\text{ Hz}$) H_1 ; 3.55 (3H, s) OCH_3 ; 3.50-3.35 (2H, m) H_2 and H_3 ; 1.16, 1.13 and 1.02 (9H, 3s) 3CH_3 .

The lyxo and ribo configurations were ascribed to epoxides 9 and 10 respectively on the basis of ^1H NMR data.^{3,17}

Methyl 2,3,6 - tri - O - acetyl - 4 - deoxy - α - D - xylo - hexopyranoside (12)

To a stirred soln of 10 (0.275 g, 0.8 mmol) in 5 ml dioxane and water 1:1 (v/v) 1 ml 1 N HClO_4 was added at room Temp. After 10 days (TLC benzene-acetone 8:2) the mixture was neutralized with NaOH aq. and evaporated to dryness. The residue was treated with Ac_2O and pyridine. The soln was kept at room temp. for 1 day, poured into 10 ml of ice-water, and extracted several times with CHCl_3 . The combined chloroform extract was washed with water, 10% NaHCO_3 aq. and water, dried (MgSO_4) and evaporated to dryness. The residue was chromatographed over silica gel (5 g) with a mixture of benzene and acetone 95:5 (v/v). Compound 11 0.2 g (61%) was obtained as a colourless oil. This compound was heated for 2 hr with 5 ml 2 N KOH aq. The soln was neutralized with Amberlite IR-45 (H^+) evaporated to dryness and treated with Ac_2O and pyridine. The crude product obtained after acetylation was chromatographed over silica gel (5 g) with ligroin 60-80 $^{\circ}$ -EtOAc 8:2. The isolated compound was finally purified by crystallization from a mixture of *n*-hexane and ether, giving 72 mg (24%) of 12 as colourless crystals, m.p. 74° , $[\alpha]_{D}^{25} + 139.2 \pm 1^{\circ}$, $[\alpha]_{D}^{25} + 153 \pm 1^{\circ}$ (c, 0.56 in CHCl_3); Ref. 14 m.p. $75-76^{\circ}$, $[\alpha]_{D}^{25} + 138^{\circ}$ (CHCl_3); Ref. 18 m.p. 74° , $[\alpha]_{D}^{25} + 135.2^{\circ}$ (CHCl_3).

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